

SSD: **BIO10**

Academic year: **2017/2018**

Faculty: **Department of Chemistry**

Study courses: **Biochemistry**

Study plans/Curricula: **Chemistry 1<sup>st</sup> level degree**

Type: **???????**

Total Credits: **6**

Didactic Methods: **Lectures and laboratory**

Didactic Period: **3<sup>o</sup> year - 2<sup>o</sup> semester (March – June)**

Exam type: **Oral**

Professor in charge: **Marina Roberti**

### **Training objectives**

Acquisition of knowledge on basic aspects of biochemistry such as: structure and function of proteins and biological membranes, general bioenergetic principles, fundamental metabolic strategies and major metabolic pathways. Acquiring theoretical-practical knowledge on some biochemical techniques. Acquisition of theoretical tools for the interpretation of biochemical processes.

### **Prerequisites**

Basic knowledge of General Chemistry, Inorganic and Organic Chemistry

### **Didactic Methods**

Classroom lectures with the help of PowerPoint presentations

### **Course programme**

#### Lectures (40 h):

The molecular logic of living organisms. The Biomolecules.

Amino acids: chemical-physical properties. Proteins: primary structure and its determination methods. Secondary, tertiary and quaternary structure of proteins. Protein Purification.

Lipids: chemical-physical properties and classification. The biological membranes.

Relationships between protein structure and function: myoglobin and hemoglobin.

Enzymes: Nature, property, classification. Enzyme kinetics: meaning of  $K_m$ ,  $V_{max}$ ,  $K_{cat}$ ; graph of Lineaweaver-Burk. Enzyme inhibition: Competitive, incompetitive and non-competitive. Factors that affect enzymatic activity. Enzyme regulation.

Allosteric enzymes: molecular models. Isozymes. Main vitamins and coenzymes.

Enzymatic Dosage: direct and indirect methods of quantization of substrates and enzymatic activity.

Principle of bioenergetics: production, conservation and use of metabolic energy. General metabolic concepts and drawings. Transfer of phosphoric groups and ATP. ATP hydrolysis free energy.

Carbohydrate metabolism: aerobic and anaerobic glycolysis and its regulation. Glycogenolysis and glycogen synthesis and their regulation. Gluconeogenesis. Pentose phosphate cycle.

Lipid metabolism: Oxidation of saturated and unsaturated fatty acids of even and odd numbers of carbon atoms. Metabolism of ketone bodies. Fatty acid biosynthesis.

Amino Acid metabolism: amino acid reactions: transamination. Ammonia metabolism: ammonia generation, liver transplantation, urea cycle.

Terminal metabolism: structure and organization of mitochondria. Mitochondrial transport systems. Oxidative decarboxylation of pyruvic acid. Citric acid cycle. Mitochondrial electron transport chain: respiratory complexes, potential redox and transfer of reducing equivalents, inhibitors. Oxidative phosphorylation and chemiosmotic theory; uncoupling molecules. P/O ratio and respiratory control index. Energy yield of glucose and fatty acid oxidation.

Hints on the mechanisms of transfer of genetic information

Principles of some biochemical techniques.

#### Laboratory ( 15 h):

Practical introduction to the biochemical laboratory, protein dosage, substrate and enzymatic activity dosage, electrophoresis in denaturing and non-denaturing conditions

#### **Reference Texts**

Principi di Biochimica di Lehninger- Zanichelli